

REPORT ON RADIONUCLIDE BINDING IN CEMENT SYSTEMS

SLOW PROCESSES IN CLOSE-TO-EQUILIBRIUM CONDITIONS FOR RADIONUCLIDES IN WATER/SOLID SYSTEMS OF RELEVANCE TO NUCLEAR WASTE MANAGEMENT

SKIN

DELIVERABLE D2.4

COLLABORATIVE PROJECT (CP)

Grant agreement N°.: FP7-269688

Submitting organizations: LU

Authors: John Hinchliff, Nick Evans, Monica Felipe Sotelo

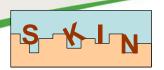
Due date of deliverable: Project Month 34 Actual submission: Project Month 37

Start date of the project: 01 January 2011

Duration: 36 months

Project co-funded by the European Commission under the Seventh Framework Programme of		
the European Atomic Energy Community (Euratom) for nuclear research and training activities		
(2007 to 2011)		
Dissemination Level		
PU	Public	×
RE	Restricted to a group specified by the partners of the project	
CO	Confidential, only for partners of the project	





REPORT ON RADIONUCLIDE BINDING IN CEMENT SYSTEMS

John Hinchliff*, Nick Evans, Monica Felipe Sotelo.

Loughborough University (UK)

* Corresponding author: cmjh5@lboro.ac.uk

Abstract

A series of radial diffusion and advection experiments have been undertaken and examples are presented where significant results have been obtained. In the diffusion experiments the effects of high ionic strength and cellulose degradation products (CDP) on the mobility of strontium are described at both tracer and tracer plus carrier concentrations. Ca migration in the same radial configuration has been observed using autoradiography. The advection experiments required the development and manufacture of a purpose designed cell, the new cell is described and the results for ⁹⁰Sr and ⁴⁵Ca are presented. All experiments have been undertaken on intact samples of the cementitious medium.

Introduction

This paper describes examples from a series of diffusion and advection experiments. These dynamic experiments aim to understand the interaction between cementitious media and radionuclides relevant to the geological disposal of radioactive waste.

The cementitious media being studied are NRVB (Nirex Reference Vault Backfill) and a waste packaging grout containing PFA (pulverised fuel ash). The experiments on the PFA grout have not yielded any evidence of diffusion and it has been assumed that precipitation has occurred due to very low solubility. Consequently, the discussion here concentrates on the NRVB experiments. The radionuclides being studied are 90 Sr, 45 Ca, 241 Am, 152 Eu and 75 Se. Migration of 90 Sr and 45 Ca has been observed in the diffusion and advection experiments but there is currently no evidence of migration of 241 Am and 152 Eu. The 90 Sr experiments ran smoothly with manageable durations (stabilising at <200 days) and as a consequence it was also possible to undertake experiments demonstrating the effects of high ionic strength and the presence of CDP on Sr mobility.





The ⁷⁵Se diffusion experiments have been running for six months and migration has not been observed.

Diffusion Experiments

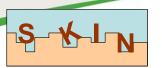
The radial diffusion experimental technique uses small pre-cast cylinders of the cementitious matrix under investigation, in this case, NRVB. The appropriate concentration of radionuclide tracer and carrier (if required) is introduced into a cavity in the centre of the cylinder which is then sealed and placed in a solution previously equilibrated with the solid matrix. The increase in concentration of the isotope in the external solution is then determined at defined time intervals.

⁹⁰Sr Diffusion</sup>

Tracer only (13.5 kBq of ⁹⁰Sr) and tracer with carrier (13.5 kBq of ⁹⁰Sr with 21.8 mg Sr as Sr(NO₃)₂; equivalent to ~10⁻³ mol dm⁻³ if fully equilibrated with the receiving solution) radial diffusion experiments on NRVB were undertaken using ⁹⁰Sr in the presence and absence of CDP. Two similar tracer only experiments were undertaken where the ionic strength of the solution was increased. In the first of these experiments sufficient Na/KCl was added to produce a 0.1 mol dm⁻³ solution. In the second experiment the same mass of Na/KCl was added to the central core with the tracer, the purpose was to create a high ionic strength gradient that would equilibrate to 0.1 mol dm⁻³ as the experiment progressed. These experiments were originally devised because the ionic strength of the CDP solutions was noted to be higher (most likely due to the 80°C production temperature) than the NRVB equilibrated water.

The ⁹⁰Sr diffusion experiments proceeded fast enough to be monitored via the increase of concentration in the surrounding solution. Figure 1 below show the results.





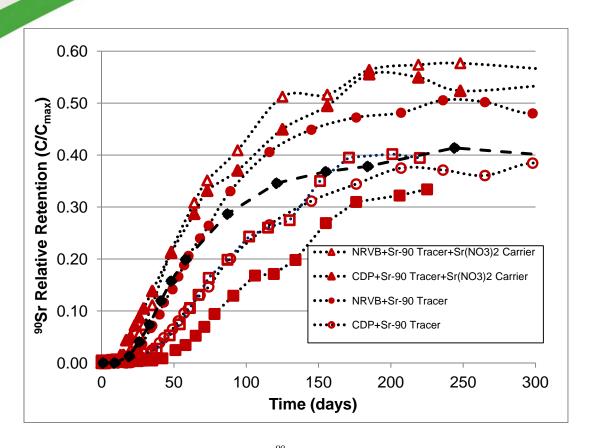


Figure.1 Results of the ⁹⁰Sr diffusion experiments

The effect on the migration of ⁹⁰Sr caused by the presence of CDP is seen to be significant at tracer concentrations where, contrary to expectations, migration was significantly slowed and retention on the NRVB increased. The effects were not evident in the carrier experiments where the Sr concentration was increased with non-active Sr(NO₃)₂. The high ionic strength tracer only experiment produced very similar results to the CDP tracer only experiment. The gradient ionic strength experiment produced the slowest initial breakthrough and subsequent diffusion. However the gradient ionic strength results are still increasing and may converge with the high ionic strength experiment.

A tracer only experiment using gluconic acid (in the lactone form, i.e. with no counter ions) as a surrogate for the CDP was also undertaken. The results, shown in black on





fig. 1, initially resembled the carrier experiments but upon stabilisation the retention on the NRVB was similar to the high ionic strength and CDP tracer only experiments.

The results show that there is an effect on ⁹⁰Sr migration at tracer concentrations due to the presence of CDP and/or the increased ionic strength. At this stage there is insufficient evidence to confirm which effect is dominant under any given set of experimental conditions.

The slowing of Sr migration could be due to the formation of a ternary complex between a component of the CDP, the surface of the cement matrix and Sr, or enhanced ion exchange with calcium in the cement phases in the matrix. Increasing electrostatic interaction is believed to be the reason why diffusion is slowed as the ionic strength increases.

Ca-45 Diffusion

There had been an expectation to observe some mobility of calcium in the cementitious systems being investigated primarily because calcium is present in high concentration in both solid (> 25% w/w) and solution (\sim 800 ppm or \sim 2x10⁻² mol dm⁻³). The addition of ⁴⁵Ca was so small in comparison that isotope exchange would be the only effect observed. However, it is not clear how migration would be driven by isotope exchange and the mechanism may simply be slow diffusion across a very small ⁴⁵Ca concentration gradient.

The short half-live of ⁴⁵Ca (163 days) could also be an issue if mobility was slow. After one year, and with no breakthrough of ⁴⁵Ca, one of the NRVB experiments was stopped and the cylinder sectioned for autoradiography. This was done because of the need to establish whether the decay of ⁴⁵Ca had rendered it difficult to detect on the autoradiography plates. The resulting images are shown below.





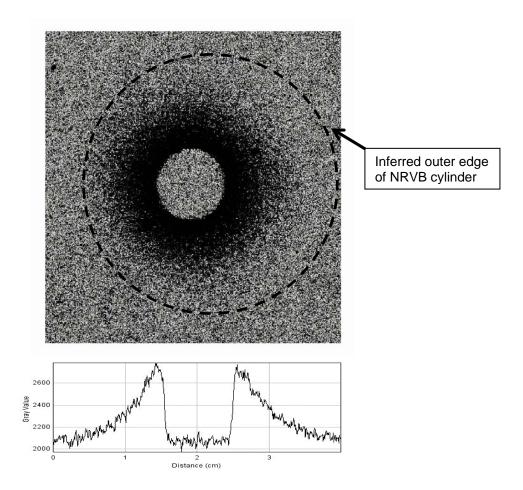


Fig. 2 Autoradiograph and intensity plot of NRVB cylinder from the ⁴⁵Ca diffusion experiments (central core "plugged" to shield highest activity at inner walls).

It is clear that ⁴⁵Ca has moved into the NRVB matrix from the central core, penetration appears to be several millimetres. The image of the activity in the matrix has been made clearer by screening, with a plastic plug (absorbing the beta radiation from ⁴⁵Ca decay), the much higher activity present on the surface of the core. The unshielded autoradiograph and intensity plot are shown below.





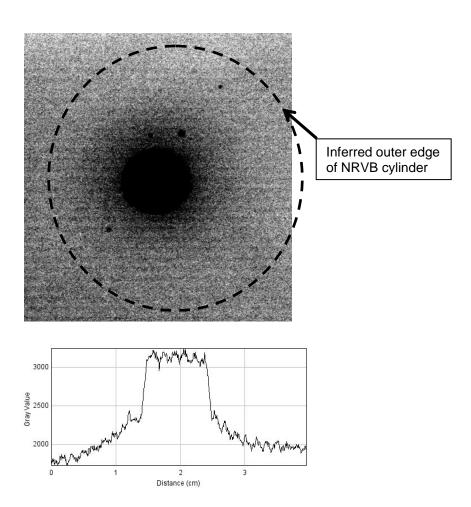


Fig. 3 Autoradiograph and intensity plot of NRVB cylinder from the Ca-45 diffusion experiments (central core "unshielded" to show highest activity at inner walls)

It should be noted that the intensity plots are not calibrated and it is only possible to infer relative concentrations. The unshielded figures demonstrate that a significant proportion of the ⁴⁵Ca has remained on the inner walls of the central core. Additionally the 2 cm³ of solution remaining in the core when the experiment was halted, was removed and analysed by liquid scintillation counting and the ⁴⁵Ca activity concentration was found to be at background.





Advection Experiments

A radial advection apparatus has been designed and manufactured and is now operational. Results for 90 Sr and 45 Ca in the presence and absence of CDP have been obtained. The photographs below show the main parts dismantled and the completed set up. The apparatus has been manufactured to enable testing of cementitious cylinders with similar dimensions those used in the diffusion experiments. The "eluent" is pushed from the steel reservoir through the cylinder using N_2 pressure. The whole system is effectively closed to O_2 and CO_2 ingress up to the end of the sample collection tube where interaction with the atmosphere is limited by the small internal diameter of the teflon tubing.



Fig.4 Photographs of the advection apparatus





The 90 Sr and 45 Ca plots below show that the method works. Flow rate control using N_2 pressure remains an issue as outgassing is visible within the cell at the NRVB surface. The "step" changes and data gaps observed in the results were generally associated with interventions e.g. refilling the reservoir, changing the gas bottle or clearing blocked tubing.

Ca-45 Advection

Figures 5 and 6 below show the results of the ⁴⁵Ca advection experiment using NRVB equilibrated water.

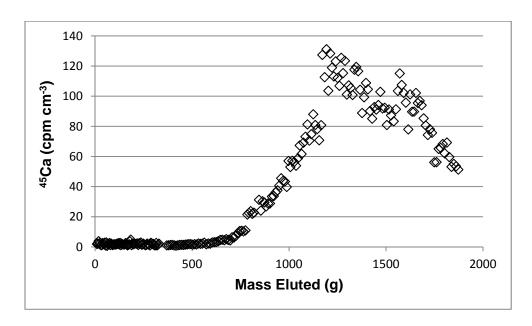
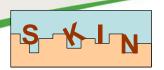


Fig.5 Graph showing advection of ⁴⁵Ca through NRVB





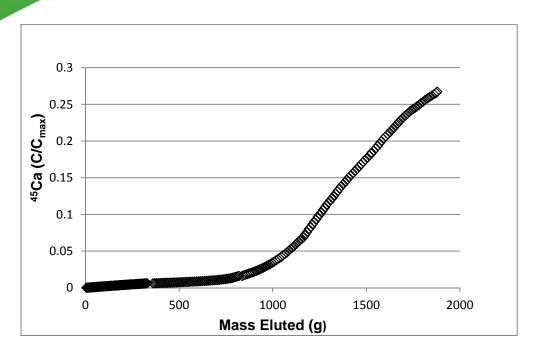
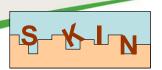


Fig.6 Graph showing cumulative ⁴⁵Ca results

The recovery of ⁴⁵Ca was less than 30% implying that over 70% was retained on the NRVB..A similar experiment which ran for the same total flow was undertaken using a CDP solution and no ⁴⁵Ca was eluted. The runs were stopped before the elution of ⁴⁵Ca had ceased so that autoradiographs could be produced, the resulting images are shown as figures 7 and 8.





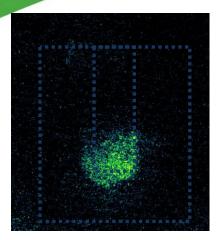


Fig.7 Autoradiograph of NRVB cylinder showing ⁴⁵Ca distribution (no CDP present)

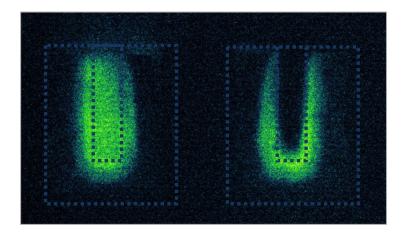


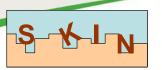
Fig.8 Autoradiograph of NRVB cylinder showing ⁴⁵Ca distribution (CDP present)unshielded image on the left, clay shielding placed in central well on right hand image.

The autoradiographs clearly show that ⁴⁵Ca is retained on the NRVB much more significantly in the presence of CDP. There was a delay after ending the experiment prior to producing the image shown as figure 7 and further decay of the ⁴⁵Ca will have occurred. As a consequence the ⁴⁵Ca intensities shown in figures 7 and 8 are not directly comparable.

Sr-90 Advection

Figures 9 and 10 below show the results of the ⁹⁰Sr advection experiments. Runs in the absence of CDP and in the presence of CDP solution are shown. Figure 10 is a cumulative plot of the data shown in figure 9. Both runs were stopped before the elution of ⁹⁰Sr had ceased so that autoradiographs could be produced.





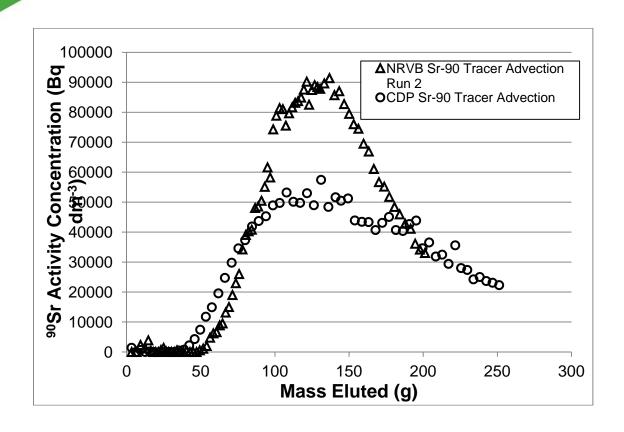
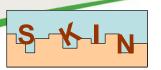


Figure 9 Graph showing results of the advection of ⁹⁰Sr through NRVB in the presence and absence of CDP





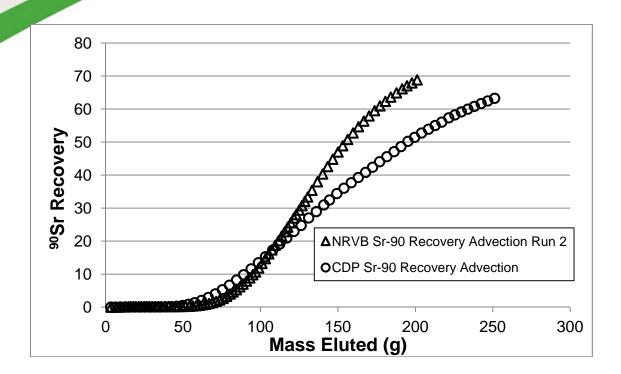


Figure 10 Graph showing cumulative results of the advection of ⁹⁰Sr through NRVB in the presence and absence of CDP

There are clear differences in the elution profiles, the most obvious being the reduced peak height of the CDP experiment. The same mass of tracer was added to each run and there is an indication that more ⁹⁰Sr is retained in the presence of CDP. However, it is also clear that the CDP profile has a longer tail, implying that all the ⁹⁰Sr will eventually elute.

The autoradiographs are shown below.





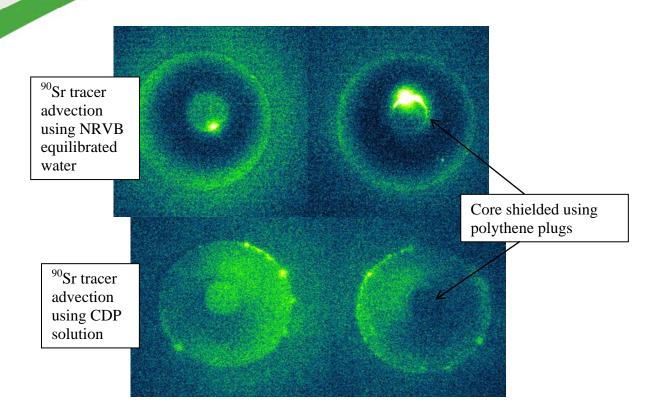
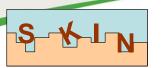


Figure 11 Autoradiographs from the ⁹⁰Sr tracer advection experiments

NRVB equilibrated water

- There is a distinct ring of ⁹⁰Sr tracer activity within the NRVB matrix extending to the outer edge.
- Spots of activity are absent from the outer edge.
- There is a ring of inactivity surrounding the central core.
- The central core has significant residual activity present, this is particularly
 evident on the upper right image where the intense activity can be clearly seen
 despite the shielding (Note that this cylinder has had two active runs).





- There is a possibility that the outer ring is Sr and calcite co-precipitation caused by the ingress of atmospheric carbon dioxide similar to that observed in the trial advections.
- The outer ring of activity could also be evidence of the tracer tail being eluted.

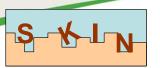
CDP solution

- More tracer is present in the NRVB matrix than seen in the NRVB equilibrated water experiment.
- The tracer is more evenly distributed than in the absence of CDP although there are distinct areas where more tracer is present, suggesting preferential flow or an artefact associated with the position of the tracer injection.
- There are spots of tracer activity on the outer edge which appear to be contiguous with the areas of possible preferential flow.
- There is residual tracer activity in the central core but none of the very intense spots seen in the central core of the NRVB equilibrated water experiment.
- The ring of activity around the outer edge is present though not as extensive as in the NRVB equilibrated water experiment.
- There is a possibility that the outer ring is Sr and calcite co-precipitation caused by the ingress of atmospheric carbon dioxide similar to that observed in the trial advections.

Advection – concluding remarks

The advection experiments show that there are clear differences in mobility between the radionuclides tested. The comparison between ⁹⁰Sr and ⁴⁵Ca is significant. The ⁴⁵Ca results indicate a strong interaction with the NRVB which is most likely due to a combination of isotope exchange and solubility limitation. The ⁴⁵Ca results in the presence of CDP may indicate that precipitation/dissolution of an organic Ca salt is the





dominant effect. Additional work will be required to confirm this. The ⁹⁰Sr results indicate that it is more mobile than ⁴⁵Ca and has a less significant interaction with the solid matrix.

Acknowledgement

The research leading to these results has received funding from the European Union's European Atomic Energy Community's (Euratom) Seventh Framework Programme FP7/2007-2011 under grant agreement n° 295722 (FIRST-Nuclides project).

References

Abdullah M. Alshamsi and Hassan D. A. Imran; *Development of a permeability apparatus for concrete and mortar*. Cement and Concrete Research Volume 32, Issue 6, June 2002, Pages 923-929.

Evans N D M. *Binding mechanisms of radionuclides to cement*. Cement and Concrete Research Volume 38, Issue 4, April 2008, pages 543-553.

Rowe P W, Barden L; *A new consolidation cell*. Géotechnique, Vol. 16, No. 2, 1966, pages 116-124.

Francis, A J, Cather, R, and Crossland, I G, *Development of the Nirex Reference Vault Backfill; report on current status in 1994*, Nirex Science Report S/97/014, UK Nirex Ltd., Harwell, UK, 1997.

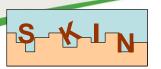
Jakob A, *Diffusion of tritiated water (HTO) and 22Na+-ions through non-degraded hardened cement pastes – II. Modelling results.* PSI-Bericht Nr. 02-21 December 2002. Markovaara-Koivisto, M., Read, D., Lindberg, A., Siitari-Kauppi, M., Togneri, L., 2009. Geology of the Sievi, Kuru and Askola sites. Uranium mineralogy at Askola. Research Report TKK-GT-A-4, Helsinki University of Tecnology.

Andersson K, Torstenfelt B. Allard B: *Sorption and diffusion studies of Cs and I in concrete* Department of Nuclear Chemistry Chalmers University of Technology Göteborg, Sweden January 1983

Mibus J, Sachs S, Pfingsten W, Nebelung C, Bernhard G, *Migration of uranium(IV)/(VI) in the presence of humic acids in quartz sand*: A laboratory column study. Journal of Contaminant Hydrology 89 (2007).

M.M. Askarieh, A.V. Chambers, F.B.D. Daniel, P.L. FitzGerald, G.J. Holtom, N.J. Pilkington, J.H. Rees. *The chemical and microbial degradation of cellulose in the near field of a repository for radioactive wastes*. Waste Management 20 2000 pp 93-106.





Pala´gyi S, Tamberg K S, Vodickova H; *Transport and sorption of Sr-85 and I-125 in crushed crystalline rocks under dynamic flow conitions*. Journal of Radioanalytical Nuclear Chemistry 2010 page 283

Tits J, Wieland R, Bradbury M H; *The effect of isosaccharinic acid and gluconic acid on the retention of Eu(III), Am(III) and Th(IV) by calcite.* Applied Geochemistry 20 November 2005 Volume 20, Issue 11 pages 2082-2096

Chatterji, S., 1994. Transportation of ions through cement based materials. Part 1: Fundamental equations and basic measurement techniques. Cem. Concr. Res. 24, 907-912.

Chatterji, S., 1995. On the applicability of Fick's second law to chloride ion migration through Portland cement concrete. Cem. Concr. Res. 25, 299-303.

Chatterji, S., 1999. Evidence of variable diffusivity of ions in saturated cementitious materials. Cem. Concr. Res. 29, 595-598.